



Absolute Stereostructure of Potent α -Glucosidase Inhibitor, Salacinol, with Unique Thiosugar Sulfonium Sulfate Inner Salt Structure from *Salacia reticulata*

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Abstract—A most potent α -glucosidase inhibitor named salacinol has been isolated from an antidiabetic Ayurvedic traditional medicine, *Salacia reticulata* WIGHT, through bioassay-guided separation. The absolute stereostructure of salacinol was determined on the basis of chemical and physicochemical evidence, which included the alkaline degradation of salacinol to 1-deoxy-4-thio-D-arabinofuranose and the X-ray crystallographic analysis, to be the unique spiro-like configuration of the inner salt comprised of 1-deoxy-4-thio-D-arabinofuranosyl sulfonium cation and 1'-deoxy-D-erythrosyl-3'-sulfate anion. Salacinol showed potent inhibitory activities on several α -glucosidases, such as maltase, sucrase, and isomaltase, and the inhibitory effects on serum glucose levels in maltose- and sucrose-loaded rats (in vivo) were found to be more potent than that of acarbose, a commercial α -glucosidase inhibitor. © 2002 Elsevier Science Ltd. All rights reserved.

Introduction

The Hippocrateaceae plant¹ *Salacia reticulata* WIGHT ('Kotala himbutu' in Singhalese), a large woody climbing plant in the submontane forests, is distributed in Sri Lanka and the south region of India. The roots and stems of *S. reticulata* have been extensively used for the treatments of rheumatism, gonorrhea, and skin diseases, particularly as a specific remedy at the initial stages of diabetes in the Ayurvedic system of Indian traditional medicine.² The aqueous extract from the root bark of Sri Lankan *S. reticulata* was reported to show hypoglycemic activity in rats,³ while the aqueous extract was also reported to produce a significant lowering of plasma glucose in streptozotocin-induced diabetic rats.⁴ As the constituents of this plant, many triterpenes as well as hydrocarbon, sitosterol, mangiferin, and gutta-percha were isolated from the root and stem barks.⁵ However, pharmacologically active components were left uncharacterized.

In the course of our studies on antidiabetogenic compounds from natural medicines and medicinal food-stuffs,⁶ we found that the aqueous methanolic extract of Sri Lankan *S. reticulata* did not show hypoglycemic effects on oral D-glucose-loaded rats and alloxan-induced diabetic mice, while the extract and the water-soluble portion showed potent hypoglycemic effect in oral sucrose-loaded rats and α -glucosidase inhibitory effect. Through bioassay-guided separation using α -glucosidase inhibitory activities, we isolated potent α -glucosidase inhibitors termed salacinol (**1**)⁷ and kotalanol⁸ from the water-soluble portion together with several phenolic compounds.⁹ Furthermore, **1** and kotalanol were also isolated from Indian *S. oblonga* and Thai *S. chinensis* together with several new triterpenes with aldose reductase inhibitory activity.¹⁰

This paper presents a full account of the isolation and structural elucidation of salacinol (**1**) from the roots and stems of Sri Lankan *S. reticulata*.¹¹ In addition, we describe the inhibitory properties of the extracts and **1** on the increase of serum glucose levels in several sugar-loaded rats (in vivo) and the inhibitory activities of **1** on various glucosidase (in vitro).

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Table 1. Inhibitory effects of the MeOH extract from *Salacia reticulata* on serum glucose levels in several sugar-loaded rats

Treatment	Dose (mg/kg, po)	N	Serum glucose (mg/dL)		
			0.5 h	1.0 h	2.0 h
<i>Maltose-loaded</i>					
Normal	—	12	81.3±4.2**	97.2±4.3**	94.5±4.1**
Control	—	12	212.1±6.2	183.5±8.6	129.3±2.6
MeOH ext.	50	6	160.0±11.9*	164.7±4.8	142.9±4.5
	100	6	168.8±9.2**	160.9±7.0	141.3±3.3
	200	6	116.9±6.9**	132.4±5.7**	130.3±3.1
<i>Sucrose-loaded</i>					
Normal	—	12	77.2±3.3**	94.8±5.0**	92.7±4.7**
Control	—	12	182.9±5.5	153.0±3.9	129.0±2.8
MeOH ext.	50	6	142.3±7.0**	128.4±2.7**	123.6±1.9
	100	6	123.4±9.7**	118.6±6.8**	125.2±3.7
	200	6	97.2±4.0**	105.3±4.4**	104.2±3.2**
<i>Glucose-loaded</i>					
Normal	—	12	85.9±4.0**	88.9±2.9**	93.3±3.2
Control	—	12	191.2±5.5	156.1±4.5	120.6±2.2
MeOH ext.	50	6	169.6±9.9	163.7±12.2	146.4±19.1
	100	6	197.8±9.4	179.2±11.9	148.4±18.8
	200	6	198.0±13.7	200.7±21.5**	167.1±33.9

Each value represents the mean±SEM. Significantly different from the control: * $p < 0.05$, ** $p < 0.01$.

Results and Discussion

The isolation of the chemical constituents from *S. reticulata* was carried out through the following procedure. The dried roots and stems of *S. reticulata* (collected in Sri Lanka) were extracted with methanol under reflux. The methanolic extract was found to inhibit the increase of serum glucose levels in maltose- and sucrose-loaded rats as shown in Table 1, whereas the methanolic extract did not show the inhibitory effects on serum glucose levels in glucose-loaded rats. In addition, the effect of the methanolic extract on serum glucose levels in alloxan-induced diabetic mice, which is known to be a experimental model for insulin-dependent diabetics, was examined. However, the methanolic extract lacked hypoglycemic effect at the single administration of 3000 mg/kg (po) as shown in Table 2. The methanolic extract was then partitioned into an ethyl acetate (AcOEt) and water (H₂O) mixture to give an AcOEt-soluble fraction and H₂O-soluble fraction. The H₂O-soluble fraction was found to show inhibitory effect on the increase of serum glucose levels in sucrose-loaded rats, but the AcOEt-soluble fraction did not affect at a dose of 50 mg/kg (Table 3).

The results of above-mentioned in vivo screening test led us to confirm that the mechanism of antidiabetic effects of this natural medicine was suggested to be inhibitory activity for small intestinal α -glucosidase, and the α -glucosidase inhibitory constituent was concentrated in the H₂O-soluble fraction (maltase: MeOH extract IC₅₀=42 μ g/mL, H₂O-soluble fraction IC₅₀=35 μ g/mL, sucrase: MeOH extract IC₅₀=32 μ g/mL, H₂O-soluble fraction IC₅₀=26 μ g/mL).

The isolation of the α -glucosidase inhibitor from the H₂O-soluble fraction was carried out by bioassay-guided separation using inhibition indices for maltase and sucrase from small intestinal brush border membrane

Table 2. Effects of the MeOH extract from *Salacia reticulata* on serum glucose levels in alloxan-induced diabetic mice

Treatment	Dose (mg/kg)	N	Serum glucose (mg/dL)		
			0 h	1 h	3 h
Control	—	8	689.3±29.2	604.3±15.4	580.4±10.0
MeOH ext.	250 (po)	8	684.8±26.7	565.6±20.2	616.4±18.4
Insulin	1 (U/kg, ip)	8	690.3±40.3	107.1±19.0**	107.4±8.9**
Control	—	8	625.3±19.3	527.4±31.4	533.6±18.0
MeOH ext.	1500 (po)	8	618.6±27.0	550.1±20.7	527.3±22.4
	3000 (po)	8	610.1±25.3	547.9±43.0	583.1±36.9

Each value represents the mean±SEM. Significantly different from the control: ** $p < 0.01$.

Table 3. Inhibitory effects of AcOEt- and H₂O-soluble fractions from *Salacia reticulata* on serum glucose levels in sucrose-loaded rats

Treatment	Dose (mg/kg, po)	N	Serum glucose (mg/dL)
			0.5 h
Normal	—	5	82.7±5.9**
Control	—	5	168.4±9.3
AcOEt-soluble fraction	50	5	162.0±3.7
	200	5	140.6±5.8*
H ₂ O-soluble fraction	10	6	154.4±7.0
	25	6	128.5±2.0**
	50	5	117.4±5.4**
	100	6	106.1±4.6**

Each value represents the mean±SEM. Significantly different from the control: * $p < 0.05$, ** $p < 0.01$.

vesicles (Chart 1). The H₂O-soluble fraction was suspended with methanol and then filtered through Kiri-yama funnel to give a methanol-soluble phase and a residue. The inhibitory activities were concentrated in the methanol-soluble phase (maltase: IC₅₀ = 30 µg/mL, sucrase: IC₅₀ = 18 µg/mL), so that the methanol-soluble phase was subjected to ordinary-phase silica gel chromatography (CHCl₃–MeOH–H₂O) to give eight fractions (Fr. 1–8). The active fractions [Fr. 3 and 4 (maltase: IC₅₀ = 60, 41 µg/mL, sucrase: IC₅₀ = 15, 6.7 µg/mL)] and Fr. 2 were successively subjected to ODS and NH column chromatography and finally HPLC [Shodex SC1011 (Ca²⁺), SP0811 (Pb²⁺), and YMC-Pack Polyamine II] to furnish salacinol (**1**, 0.025% from natural medicine), dulcitol (0.047%), D-glucose (0.100%), D-fructose (0.138%), sucrose (0.099%), and glycerol (0.010%).

Absolute stereostructure of salacinol (1**).** Salacinol (**1**) was isolated as colorless prisms of mp 187–189 °C (EtOH) with positive optical rotation ($[\alpha]_D^{28} +4.9^\circ$) and was deduced to possess a sulfate group by the positive potassium rhodizonate test.¹² The positive-ion fast atom bombardment (FAB)-MS of **1** showed a quasimolecular ion peaks at m/z 357 (M+Na)⁺ and 335 (M+H)⁺ in addition to a fragment ion peak at m/z 255 (M–SO₃+H)⁺, while a quisimolecular ion peak was observed at m/z 333 (M–H)[–] in the negative-ion FAB-MS. The molecular formula of **1** has been shown to be C₉H₁₈O₉S₂ by high-resolution MS analysis. The IR spectrum of **1** showed absorption bands due to hydroxyl (3417, 1073, 1019 cm^{–1}) and sulfate (1262, 1238, 801 cm^{–1}) functions. The ¹H and ¹³C NMR (Table 4) spectra of **1** showed signals assignable to four methylenes linking to a heteroatom {δ (pyridine-*d*₅): 4.33 (2H, br s,

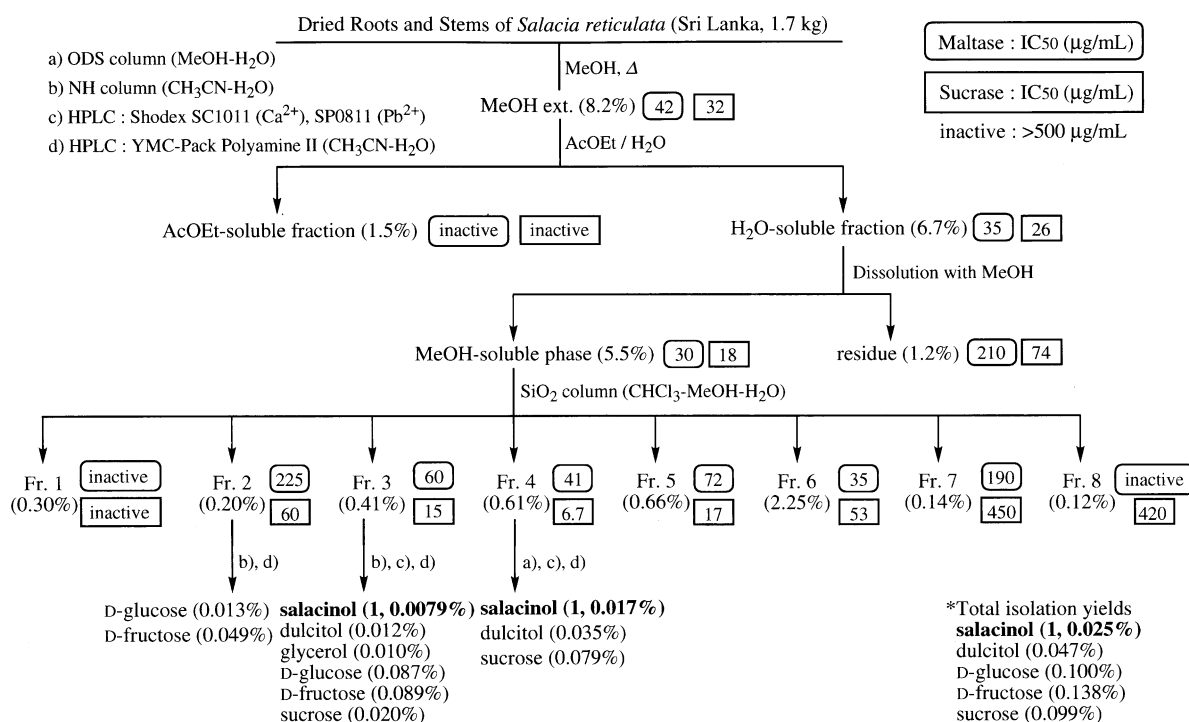


Chart 1. Bioassay-guided separation from the died roots and stems of *Salacia reticulata* using α-glucosidase inhibitory activities.

Table 4. ¹H NMR and ¹³C NMR data for salacinol (**1**)

Solvent	Pyridine- <i>d</i> ₅		CD ₃ OD		CD ₃ OH
	¹ H NMR	¹³ C NMR	¹ H NMR	¹³ C NMR	¹³ C NMR
C-1	4.33 (2H, br s)	50.7	3.83 (2H, br s)	51.44	51.45
C-2	5.10 (1H, br s)	78.5	4.60 (1H, br dd, <i>J</i> = ca. 3, 5 Hz)	79.24	79.36
C-3	5.12 (1H, dd-like)	79.0	4.40 (1H, dd, <i>J</i> = 1.2, 2.7 Hz)	79.75	79.90
C-4	4.69 (1H, t-like)	72.6	4.01 (1H, br dd, <i>J</i> = ca. 5, 7 Hz)	73.43	73.50
C-5	4.51 (1H, dd, <i>J</i> = 8.0, 11.6 Hz)	60.2	3.95 (1H, dd, <i>J</i> = 7.1, 10.5 Hz)	60.95	61.06
	4.54 (1H, dd, <i>J</i> = 6.8, 11.6 Hz)		4.03 (1H, dd, <i>J</i> = 5.2, 10.5 Hz)		
C-1'	4.62 (1H, dd, <i>J</i> = 4.2, 13.1 Hz)	52.6	3.86 (1H, dd, <i>J</i> = 6.4, 13.1 Hz)	52.35	52.38
	4.76 (1H, dd, <i>J</i> = 4.9, 13.1 Hz)		3.97 (1H, dd, <i>J</i> = 3.6, 13.1 Hz)		
C-2'	4.99 (1H, ddd, <i>J</i> = 4.2, 4.9, 7.6 Hz)	67.6	4.34 (1H, ddd, <i>J</i> = 3.6, 6.4, 7.6 Hz)	67.69	67.81
C-3'	5.25 (1H, ddd, <i>J</i> = 3.7, 3.9, 7.6 Hz)	79.4	4.29 (1H, ddd, <i>J</i> = 3.4, 3.5, 7.6 Hz)	80.56	80.59
C-4'	4.37 (1H, dd, <i>J</i> = 3.9, 11.6 Hz)	62.2	3.83 (1H, dd, <i>J</i> = 3.4, 12.2 Hz)	61.75	61.88
	4.60 (1H, dd, <i>J</i> = 3.7, 11.6 Hz)		3.94 (1H, dd, <i>J</i> = 3.5, 12.2 Hz)		

¹H NMR measured at 500 MHz. ¹³C NMR measured at 125 MHz.

1-H₂), [4.37 (dd, $J=3.9$, 11.6 Hz), 4.60 (dd, $J=3.7$, 11.6 Hz), 4'-H₂], [4.51 (dd, $J=8.0$, 11.6 Hz), 4.54 (dd, $J=6.8$, 11.6 Hz), 5-H₂], [4.62 (dd, $J=4.2$, 13.1 Hz), 4.76 (dd, $J=4.9$, 13.1 Hz), 1'-H₂}] and five methines bonding to a heteroatom [δ (pyridine-*d*₅): 4.69 (t-like, 4-H), 4.99 (ddd, $J=4.2$, 4.9, 7.6 Hz, 2'-H), 5.10 (br s, 2-H), 5.12 (dd-like, 3-H), 5.25 (ddd, $J=3.7$, 3.9, 7.6 Hz, 3'-H)]. The proton and carbon signals in the ¹H and ¹³C NMR

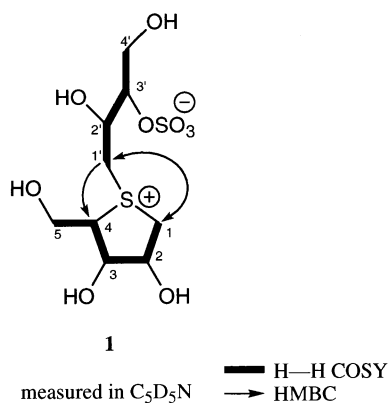


Figure 1. H—H COSY and HMBC correlations of salacinol (1).

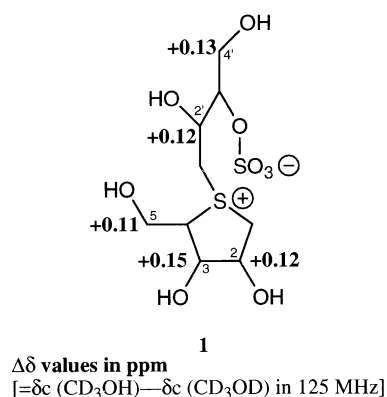


Figure 2. Application of deuterium shift rule for salacinol (1).

spectra of **1** were completely assigned with the aid of disortionless enhancement by polarization transfer (DEPT), homo- and hetero-correlation spectroscopy (¹H–¹H and ¹³C–¹H COSY), heteronuclear multiple bond correlation (HMBC), and homo- and heteronuclear Hartmann–Hahn spectroscopy (¹H–¹H and ¹³C–¹H HOHAHA) experiments and the planar structure of **1** was elucidated. Thus, the H–H COSY experiment on **1** indicated the presence of two partial structures (1-C–5-C and 1'-C–4'-C) written in bold line as shown in Figure 1. In the HMBC experiment, long-range correlations were observed between the 1'-proton and the 4-carbon, between the 1-proton and 1'-carbon, and between the 1'-proton and 1-carbon, respectively (Fig. 1).

In order to facilitate the locations of free hydroxyl groups in **1**, we compared the ¹³C NMR spectra taken in CD₃OH and CD₃OD. The deuterium shifts were observed at five carbon signals due to the 2, 3, 5, 2', and 4' positions by 0.11–0.15 ppm, while the signal due to the sulfate-bearing carbon (3'-C) was slightly changed as shown in Figure 2.

Next, the relative stereostructure of **1** was clarified by the nuclear Overhauser enhancement spectroscopy (NOESY) experiment and X-ray crystallographic analysis. Namely, nuclear Overhauser effect (NOE) was observed between the 2-proton and the 4-proton. As shown in X-ray molecular structure (computer-generated perspective drawing), the molecular conformation of **1** was found to show the unique spiro-like configuration of the inner salt comprised of 1-deoxy-4-thioarabinofuranosyl sulfonium cation and 1'-deoxyerythrosyl-3'-sulfate anion (Fig. 3). Finally, alkaline treatment of **1** with 1% sodium methoxide (NaOMe)-methanol liberated 1-deoxy-4-thio-D-arabinofuranose (**1a**), which was identified with an authentic sample prepared from D-xylose using the previously reported method.¹³ Consequently, the absolute stereostructure of salacinol (**1**) was determined as shown.

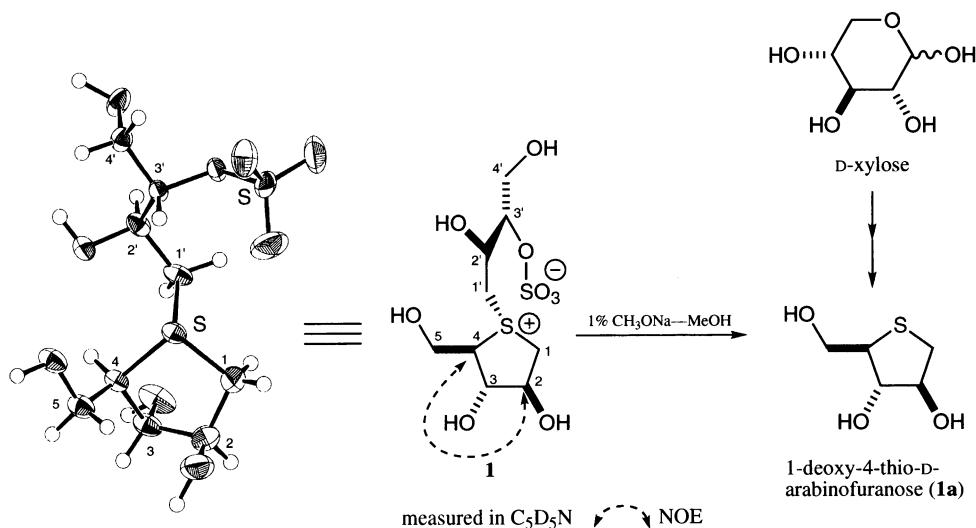


Figure 3. Absolute stereostructure and conformation of salacinol (1).

α -Glucosidase inhibitory activities of salacinol (1) and related compounds (1b, 1c). Salacinol (1) showed the competitive inhibition for the small intestinal α -glucosidase in vitro and other constituents such as dulcitol did not exhibit the activity. Table 5 indicated 50% inhibitory concentration (IC_{50}) values of 1 and commercial medicines as α -glucosidase inhibitors, acarbose and voglibose. The maltase inhibitory activity was slightly weaker than those of acarbose and voglibose (1: IC_{50} = 3.2 μ g/mL; acarbose: 1.3 μ g/mL; voglibose: 0.32 μ g/mL) and the sucrase inhibitory activity was similar to that of acarbose (1: 0.84 μ g/mL; acarbose: 1.1 μ g/mL), while the isomaltase inhibitory activity was more potent than that of acarbose and equivalent to that of voglibose (1: 0.59 μ g/mL; acarbose: 100 μ g/mL; voglibose: 0.56 μ g/mL).

On the other hand, 1-deoxy-4-thio-D-arabinofuranose (1a) showed no activity (IC_{50} : maltase >400 μ g/mL, sucrase >400 μ g/mL). This evidence suggested that the

side chain having sulfate anion was essential for the activity.

In order to examine the types of inhibitions of maltase, sucrase, and isomaltase by salacinol (1), the small intestinal brush border membrane vesicles were incubated with increasing concentrations of maltose (3.1–37 mM), sucrose (4.6–37 mM), and isomaltose (0.46–3.7 mM). The results plotted according to Lineweaver–Burk revealed a fully competitive type of inhibition on each α -glucosidase and K_i values of 1 on maltase, sucrase, and isomaltase were 0.31, 0.32, and 0.47 μ g/mL, respectively (Fig. 4 and Table 6).

Inhibitory effects of salacinol (1) on serum glucose levels in maltose- and sucrose-loaded rats. The effects of salacinol (1) on increase of serum glucose levels in maltose- and sucrose-loaded rats were examined. As shown in Table 7, 1 showed stronger inhibition of the increase of serum glucose levels in maltose- and sucrose-loaded rats

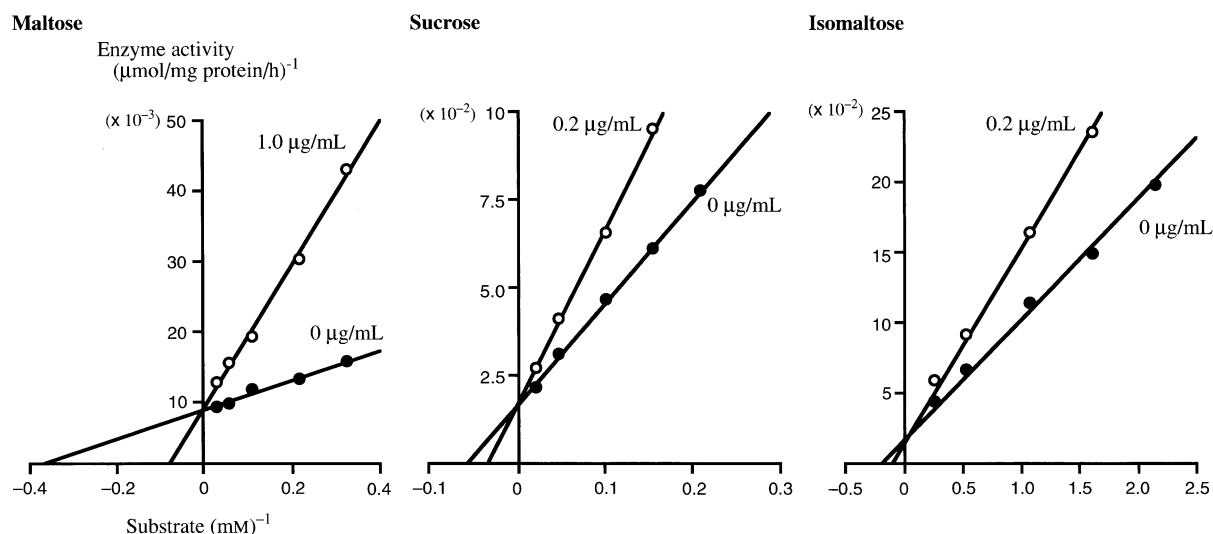


Figure 4. Lineweaver–Burk plots of the inhibition of rat intestinal maltase, sucrase, and isomaltase activities by salacinol (1).

Table 5. IC_{50} values of salacinol (1), 1a, acarbose, and voglibose for rat small intestinal disaccharidase and β -glucosidase

Enzyme (Substrate concentration)	IC_{50} (μ g/mL)			
	Salacinol (1)	1a	Acarbose	Voglibose
Maltase (37 mM)	3.2	>400	1.3	0.32
Sucrase (37 mM)	0.84	>400	1.1	0.059
Isomaltase (3.7 mM)	0.59	—	100	0.56
Trehalase (3.7 mM)	>400	—	>400	190
β -Glucosidase (2.5 mM) ^a	>400	—	>400	>400

^a*p*-Nitrophenyl- β -D-glucopyranoside.

Table 6. K_i values of salacinol (1), acarbose, and voglibose for rat small intestinal disaccharidase

Substrate	K_m (M)	K_i (μ g/mL)		
		Salacinol (1)	Acarbose	Voglibose
Maltose	2.7×10^{-3}	0.31	0.12	0.032
Sucrose	2.0×10^{-2}	0.32	0.37	0.018
Isomaltose	4.5×10^{-3}	0.47	75	0.41

Table 7. Inhibitory effects of salacinol (**1**) and acarbose on serum glucose levels in maltose- and sucrose-loaded rats

Treatment	Dose (mg/kg, po)	N	Serum glucose (mg/dL)		
			0.5 h	1.0 h	2.0 h
<i>Maltose-loaded</i>					
Normal	—	5	56.8±4.3**	60.1±2.8**	65.4±4.4*
Control	—	6	197.8±8.8	134.5±7.5	90.6±4.8
Salacinol (1)	5	6	175.5±12.4	136.0±5.4	109.1±6.1
	10	6	171.0±5.3	129.6±6.6	100.3±5.3
	25	4	137.7±7.6**	126.8±6.1	101.4±8.9
Acarbose	25	6	175.6±9.0	138.7±8.1	100.3±5.6
	50	6	173.9±8.8	137.9±3.8	104.8±4.3
	100	6	163.0±6.0	132.4±5.3	102.0±5.2
	200	6	135.6±12.6**	117.6±6.4	103.8±6.1
<i>Sucrose-loaded</i>					
Normal	—	5	66.0±4.1**	74.0±3.4**	78.4±2.0**
Control	—	7	174.6±5.1	149.7±5.1	125.1±4.6
Salacinol (1)	1.25	5	128.0±10.0**	118.4±4.9**	115.0±3.8
	2.5	5	126.4±3.2**	122.4±2.0**	121.8±2.7
	5	4	102.6±3.8**	105.8±3.9**	102.8±2.9**
Normal	—	5	74.6±3.4**	89.4±4.4**	79.7±6.6
Control	—	5	153.1±4.6	143.5±6.1	95.0±3.3
Acarbose	1	5	150.9±9.4	146.6±4.7	93.4±3.6
	2	5	141.2±6.2	152.5±7.1	106.3±3.6
	5	5	126.1±6.9*	137.2±8.3	113.8±7.2
	10	5	100.8±5.2**	118.2±2.9*	96.8±3.1

Each value represents the mean±SEM. Significantly different from the control: * $p < 0.05$, ** $p < 0.01$.

than acarbose. Therefore, it is concluded that **1** is a potent α -glucosidase inhibitor isolated from this natural medicine and is a responsible constituent of the anti-diabetic Ayurvedic traditional medicine 'Kotala himbutu', the roots and stems of *S. reticulata*.

Experimental

The following instruments were used to obtain physical data: specific rotations, Horiba SEPA-300 digital polarimeter ($l = 5$ cm); IR spectra, Shimadzu FTIR-8100 spectrometer; MS and high-resolution MS, JEOL JMS-GCMATE mass spectrometer; X-ray crystallographic analysis, Rigaku AFC5R diffractometer; ^1H NMR spectra, JNM-LA500 (500 MHz) spectrometer; ^{13}C NMR spectra, JNM-LA500 (125 MHz) spectrometers with tetramethylsilane as an internal standard; HPLC detector, Shimadzu RID-6A refractive index detector.

The following experimental conditions were used for chromatography: ordinary-phase column chromatography; Silica gel BW-200 (Fuji Silysia Chemical, Ltd., 150–350 mesh), reversed-phase silica gel column chromatography; Chromatorex ODS DM1020T (Fuji Silysia Chemical, Ltd., 100–200 mesh); NH column chromatography, Chromatorex NH-DM1020 (Fuji Silysia Chemical, Ltd., 100–200 mesh). Packed column for HPLC: Shodex SC1011 (Ca^{2+} , 8.0×300 mm, i.d.), Shodex SP0810 (Pb^{2+} , 8.0×300 mm, i.d.), and YMC-Pack Polyamine II (4.6×250 mm, i.d. and 10×250 mm, i.d.).

TLC, pre-coated TLC plates with Silica gel 60F₂₅₄ (Merck, 0.25 mm) (normal-phase) and Silica gel RP-18 F_{254S} (Merck, 0.25 mm) (reversed-phase); HPTLC, pre-coated TLC plates with Silica gel RP-18 WF_{254S} (Merck, 0.25 mm) (reversed-phase). Detection was done by spraying with 1% $\text{Ce}(\text{SO}_4)_2$ –10% aqueous H_2SO_4 followed by heating.

Extraction and isolation of salacinol (**1**) and known compounds from the dried stems of *Salacia reticulata*

The dried roots and stems of *S. reticulata* (1.7 kg, collected in Sri Lanka) was crushed and extracted three times with MeOH under reflux for 3 h. Evaporation of the solvent under reduced pressure provided the MeOH extract (139 g, 8.2%), and a part of it (130 g) was partitioned in an AcOEt– H_2O (1:1) mixture. Removal of the solvent under reduced pressure gave the AcOEt-soluble fraction (23 g, 1.5%) and H_2O -soluble fraction (107 g, 6.7%). The H_2O -soluble fraction (50 g) was suspended with MeOH and then filtered through Kiriya funnel to give a methanol-soluble phase (41 g, 5.5%) and residue (9 g, 1.2%). The methanol-soluble phase (31 g) was subjected to ordinary-phase silica gel chromatography [1.5 kg, CHCl_3 –MeOH– H_2O (6:4:1 \rightarrow 5:5:1 \rightarrow 3:7:1) \rightarrow MeOH] to give eight fractions [Fr. 1 (1.78 g), Fr. 2 (1.22 g), Fr. 3 (2.46 g), Fr. 4 (3.69 g), Fr. 5 (3.96 g), Fr. 6 (13.50 g), Fr. 7 (0.87 g), Fr. 8 (0.74 g)]. Fraction 2 (1.00 g) was subjected to NH column chromatography [30 g, CH_3CN – H_2O (70:30, v/v) \rightarrow H_2O] to give 70% aqueous CH_3CN -eluted fraction (Fr. 2-1, 400 mg), and H_2O -eluted fraction (Fr. 2-2, 600 mg). Fraction 2-1 (400 mg)

was further purified by HPLC [YMC-Pack Polyamine II, CH₃CN–H₂O (70:30, v/v)] to furnish D-glucose (25 mg, 0.013%) and D-fructose (96 mg, 0.049%). Fraction 3 (700 mg) was subjected to NH column chromatography [20 g, CH₃CN–H₂O (70:30, v/v) → H₂O] to give 70% aqueous CH₃CN-eluted fraction (Fr. 3-1, 460 mg), and H₂O-eluted fraction (Fr. 3-2, 240 mg). Fraction 3-1 (460 mg) was further purified by HPLC [Shodex SC1011 (Ca²⁺), Shodex SP0811 (Pb²⁺), H₂O, column temperature 80 °C] and [YMC-Pack Polyamine II, CH₃CN–H₂O (70:30, v/v)] to furnish salacinol (**1**, 13 mg, 0.0079%), dulcitol (20 mg, 0.012%), glycerol (17 mg, 0.010%), D-glucose (152 mg, 0.087%), D-fructose (153 mg, 0.089%), and sucrose (34 mg, 0.020%). Fraction 4 (2.60 g) was subjected to reversed-phase silica gel column chromatography [50 g, MeOH→H₂O] to give MeOH-eluted fraction (Fr. 4-1, 1.50 g), and H₂O-eluted fraction (Fr. 4-2, 1.10 g). Fraction 4-2 (1.10 g) was further purified by HPLC [Shodex SC1011 (Ca²⁺), Shodex SP0811 (Pb²⁺), H₂O, column temperature 80 °C] and [YMC-Pack Polyamine II, CH₃CN–H₂O (70:30, v/v)] to furnish salacinol (**1**, 73 mg, 0.017%), dulcitol (149 mg, 0.035%), and sucrose (337 mg, 0.079%). Their known constituents were identified by comparison of their physical data ([α]_D, IR, ¹H NMR, ¹³C NMR) with those of commercially obtained samples.

Salacinol (**1**): Potassium rhodizonate test positive, colorless prisms from EtOH, mp 187–189 °C, [α]_D²⁸ +4.9° (c 0.35, MeOH). High-resolution positive-ion FAB-MS: calcd for C₉H₁₈O₉S₂Na (M+Na)⁺: 357.0301. Found: 357.0290. IR (KBr): 3417, 1262, 1238, 1073, 1019, 801 cm⁻¹. ¹H NMR (500 MHz, pyridine-*d*₅ and CD₃OD) δ given in Table 4. ¹³C NMR (125 MHz, pyridine-*d*₅, CD₃OD, and CD₃OH) δ c given in Table 4. Positive-ion FAB-MS: *m/z* 357 (M+Na)⁺, 335 (M+H)⁺, 255 (M–SO₃+H)⁺. Negative-ion FAB-MS: *m/z* 333 (M–H)⁻.

X-ray crystallographic analysis on salacinol (1). X-ray crystal data of salacinol (**1**) were collected by a Rigaku AFC5R diffractometer. The structure was solved by the direct method and refined with a full-matrix least-squares method.

Crystal data for salacinol (1). C₉H₁₈O₉S₂, *M* = 334.36, monoclinic, space group *P*2₁, *a* = 6.433(3), *b* = 12.927(2), *c* = 8.372(3) Å, β = 93.68(3)°, *V* = 694.8(4) Å³, *Z* = 2, μ (Mo-K α) = 4.05 cm⁻¹, *F*(000) = 352, *D*_c = 1.598 g/cm⁻³, crystal dimensions: 0.15×0.20×0.20 mm. A total of 1812 reflections (1675 unique) were collected using the ω -2 θ scan technique to a maximum 2θ value of 55°, and 1454 reflections with *I* > 3 σ (*I*) were used in the structure determination. Final *R* and *R*_w values were 0.034 and 0.038, respectively. The maximum and minimum peaks in the difference map were 0.37 and –0.24 e Å⁻³, respectively. The data have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 174565.

Alkaline hydrolysis of salacinol (1). A solution of salacinol (**1**, 15 mg, 45 μ mol) in 1% NaOMe–MeOH (2.0 mL) was stirred under reflux for 3 h. The reaction mixture was neutralized with Dowex HCR-W2 (H⁺

form) and the resin was removed by filtration. Evaporation of the solvent from the filtrate under reduced pressure gave a residue, which was purified by silica gel column chromatography [1.6 g, (CHCl₃–MeOH–H₂O, 30:3:1, lower layer) → MeOH] to furnish 1-deoxy-4-thio-D-arabinofuranose (**1a**, 3 mg, 44%). Compound **1a** was identified by comparison of the physical data ([α]_D, IR, ¹H NMR, ¹³C NMR) with authentic sample prepared from D-xylose using the previously reported method.¹³

Bioassay

Inhibitory effects on increase of serum glucose levels in oral several sugar-loaded rats. Male Wistar rats (Kiwa Laboratory Animals, Ltd., Wakayama, Japan) weighing 130–170 g were starved for 20–24 h but allowed water ad libitum. The test samples were suspended in 5% acacia solution (5 mL/kg), and then orally administered to the rats. Thirty min thereafter, a water solution (5 mL/kg) of sucrose (1.0 g/kg), maltose (1.0 g/kg), or glucose (1.0 g/kg) was orally administered. Blood (ca. 0.4 mL) was collected from infraorbital venous plexus under light ether anesthesia at 0.5, 1.0, and 2.0 h after administration of sugar and the serum glucose concentration was assayed by the glucose-oxidase method (Glucose CII-test Wako, Wako Pure Industries).

Effects on serum glucose levels in alloxan-induced diabetic mice. Male ddY mice (Kiwa Laboratory Animals, Ltd., Wakayama, Japan) weighing 25–30 g were starved for 20–24 h but allowed water ad libitum. Alloxan (50 mg/kg) in saline was injected intravenously. Two days thereafter, mice were again starved for 20–24 h, and then the methanolic extract was suspended in 5% acacia solution and given orally at 10 mL/kg. Insulin was dissolved in saline and injected intraperitoneally at 10 mL/kg. Blood was collected from infraorbital venous plexus before (0 h), and 1.0 and 3.0 h after administration of test sample.

Statistics

Statistical significance of differences was estimated by analysis of variance (ANOVA) followed by Dunnett's test.¹⁴ Values were expressed as the means \pm SEM (Tables 1–3 and 7).

Glucosidase inhibitory activity. Rat small intestinal brush border membrane vesicles were prepared¹⁵ and its suspension in 0.1 M maleate buffer (pH 6.0) was used as the small intestinal α -glucosidase of maltase, sucrase, isomaltase, and trehalase. The enzyme suspension was diluted to hydrolyse each substrate to produce D-glucose at the rate of 8.5–18.5 nmol/min/tube in the following reaction. Reaction was performed by slight modification of the procedure of Dahlqvist.¹⁶ The substrate (maltose: 37 mM, sucrose: 37 mM, isomaltose: 3.7 mM, and trehalose 3.7 mM), test compound and the enzyme in 0.1 M maleate buffer (pH 6.0, 0.2 mL) were incubated together at 37 °C. After 30 min of incubation, 0.8 mL of water was added to the test tube, and the tube was immediately immersed in boiling water for 2 min to stop the reaction,

then cooled with water. Glucose concentration was determined by the glucose-oxidase method.

To examine the effect on β -glucosidase, β -glucosidase (0.25 U/mL, from almond, Sigma) and *p*-nitrophenyl- β -D-glucopyranoside (2.5 mM) as a substrate, and test compound in 0.4 mL acetate buffer (0.1 M, pH 5.0) were incubated. After incubation for 15 min at 37°C and interrupting the reaction by addition of 0.4 mL 0.2 M sodium carbonate, the amount of *p*-nitrophenol was determined colorimetrically at 405 nm.

The IC₅₀ value was determined graphically by a plot of percent inhibition versus log of the test compound. For kinetic analyses of maltase, sucrase, and isomaltase by salacinol (**1**), the enzyme and test samples (**1**, 1.0 μ g/mL for maltase and 0.2 μ g/mL for sucrase and isomaltase; acarbose, 0.5 μ g/mL for maltase and sucrase and 50 μ g/mL for isomaltase; voglibose, 0.1 μ g/mL for maltase, 0.03 μ g/mL for sucrase, and 0.25 μ g/mL for isomaltase) were incubated with increasing concentrations of maltose (3.1–37 mM), sucrose (4.6–37 mM), or isomaltose (0.46–3.7 mM).

References and Notes

1. The family of *Salacia reticulata* is recently classified as Hippocrateaceae rather than Celastraceae.
2. Jayaweera, D. M. A. *Medicinal Plants used in Ceylon, Part 1*; National Science Council of Sri Lanka: Colombo, 1981, p 77.
3. Karunanayake, E. H.; Welihinda, J.; Sirimanne, S. R.; Sinnadorai, G. J. *Ethnopharmacol.* **1984**, *11*, 223.
4. Serasinghe, S.; Serasinghe, P.; Yamazaki, H.; Nishiguchi, K.; Hombhanje, F.; Nakanishi, S.; Sawa, K.; Hattori, M.; Namba, T. *Phytotherapy Res.* **1990**, *4*, 205.
5. (a) Kumar, V.; Wazeer, M. I. M.; Wijeratne, D. B. T. *Phytochemistry* **1985**, *24*, 2067. (b) Kumar, V.; Wijeratne, D. B. T.; Abeygunawardena, C. *Phytochemistry* **1990**, *29*, 333. (c) Gunatilaka, A. A. L.; Dhanabalasingham, B.; Karunaratne, V.; Kikuchi, T.; Tezuka, Y. *Tetrahedron* **1993**, *49*, 10397. (d) Tezuka, Y.; Kikuchi, T.; Dhanabalasingham, B.; Karunaratne, V.; Gunatilaka, A. A. L. *Nat. Prod. Lett.* **1993**, *3*, 273. (e) Tezuka, Y.; Kikuchi, T.; Dhanabalasingham, B.; Karunaratne, V.; Gunatilaka, A. A. L. *J. Nat. Prod.* **1996**, *57*, 270. (f) Dhanabalasingham, B.; Karunaratne, V.; Tezuka, Y.; Kikuchi, T.; Gunatilaka, A. A. L. *Phytochemistry* **1996**, *42*, 1377.
6. (a) Yoshikawa, M.; Murakami, T.; Harada, E.; Murakami, N.; Yamahara, J.; Matsuda, H. *Chem. Pharm. Bull.* **1996**, *44*, 1923. (b) Yoshikawa, M.; Shimada, H.; Nishida, N.; Li, Y.; Toguchida, I.; Yamahara, J.; Matsuda, H. *Chem. Pharm. Bull.* **1998**, *46*, 113. (c) Matsuda, H.; Li, Y.; Murakami, T.; Matsuura, N.; Yamahara, J.; Yoshikawa, M. *Chem. Pharm. Bull.* **1998**, *46*, 1399. (d) Yoshikawa, M.; Murakami, T.; Matsuda, H. In *Towards Natural Medicine Research in the 21st Century*; Ageta, H., Aimi, N., Ebizuka, Y., Fujita, T., Honda, G., Eds.; Elsevier: Amsterdam, 1998; p. 137. (e) Yoshikawa, M.; Murakami, T.; Li, Y.; Shimada, H.; Yamahara, J.; Matsuda, H. In *Advanced in Plant Glycosides, Chemistry and Biology*; Yang, C.-R., Tanaka, O., Eds.; Elsevier: Amsterdam, 1999; p. 19. (f) Yoshikawa, M.; Morikawa, T.; Murakami, T.; Toguchida, I.; Harima, S.; Matsuda, H. *Chem. Pharm. Bull.* **1999**, *47*, 340. (g) Yoshikawa, M.; Matsuda, H. In *Saponins in Food, Feedstuffs and Medicinal Plants*; Oleszek, W., Marston, A., Eds.; Kluwer Academic: Dordrecht, 2000; p. 189. (h) Yoshikawa, M.; Matsuda, H. *BioFactors* **2000**, *13*, 231. (i) Murakami, T.; Nakamura, J.; Kageura, T.; Matsuda, H.; Yoshikawa, M. *Chem. Pharm. Bull.* **2000**, *48*, 1720. (j) Yoshikawa, M.; Murakami, T.; Kishi, A.; Kageura, T.; Matsuda, H. *Chem. Pharm. Bull.* **2001**, *49*, 863.
7. Yoshikawa, M.; Murakami, T.; Shimada, H.; Matsuda, H.; Yamahara, J.; Tanabe, G.; Muraoka, O. *Tetrahedron Lett.* **1997**, *38*, 8367.
8. Yoshikawa, M.; Murakami, T.; Yashiro, K.; Matsuda, H. *Chem. Pharm. Bull.* **1998**, *46*, 1339.
9. Yoshikawa, M.; Nishida, N.; Shimoda, H.; Takada, M.; Kawahara, Y.; Matsuda, H. *Yakugaku Zasshi* **2001**, *121*, 371.
10. Matsuda, H.; Murakami, T.; Yashiro, K.; Yamahara, J.; Yoshikawa, M. *Chem. Pharm. Bull.* **1999**, *47*, 1725.
11. The relative stereostructure of **1** in this study was reported in our preliminary communication⁷ and the total structure including the absolute configuration was presented at the 40th Symposium on the Chemistry of Natural Products: Yoshikawa, M.; Murakami, T.; Morikawa, T.; Yashiro, K.; Matsuda, H.; Muraoka, O.; Tanabe, G.; Yamahara, J. *Symposium Paper*; Fukuoka, 1998; p. 67.
12. (a) Burma, D. P. *Ana. Chim. Acta* **1953**, *9*, 513. (b) Schneider, J. J.; Lewbart, M. L. *J. Biol. Chem.* **1956**, *222*, 787.
13. (a) Yuasa, H.; Kajimoto, T.; Wong, C.-H. *Tetrahedron Lett.* **1994**, *35*, 8243. (b) Altenbach, H.-J.; Merhof, G. F. *Tetrahedron: Asymmetry* **1996**, *7*, 3087.
14. (a) Dunnett, C. W. *J. Am. Statist. Assoc.* **1980**, *75*, 789. (b) Dunnett, C. W. *J. Am. Statist. Assoc.* **1980**, *75*, 796.
15. Kessler, M.; Acuto, O.; Storelli, C.; Murer, H.; Müller, M.; Semenza, G. *Biochim. Biophys. Acta.* **1978**, *506*, 136.
16. Dahlqvist, A. *Anal. Biochem.* **1964**, *7*, 18.